

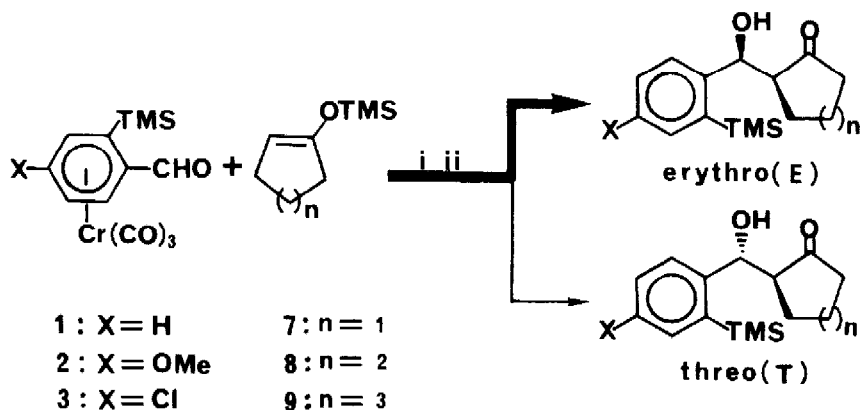
A Highly Erythro Selective Aldol Reaction Between Chromium-Complexed Benzaldehyde Derivatives and Cyclic Silyl Enol Ethers

Chisato Mukai, Won Jea Cho, and Miyoji Hanaoka*
 Faculty of Pharmaceutical Sciences, Kanazawa University
 Kanazawa 920, Japan

Summary: Reaction of chromium-complexed benzaldehydes having a trimethylsilyl group at the *ortho* position with cyclic silyl enol ethers gave, after decomplexation, the *erythro* products in a highly selective manner.

The synthetic potentiality of arene chromium tricarbonyl complexes¹⁾ has been well demonstrated. Among several significant properties, Reetz²⁾ and Uemura³⁾ has independently taken advantage of the tricarbonyl-chromium-stabilized benzylic carbocations for a stereoselective carbon-carbon bond formation. We explore here another useful application of the stabilized benzylic carbocation of arene chromium complexes in a highly stereoselective reaction. This communication deals with the successful first example of the highly *erythro* selective aldol reaction of chromium-complexed benzaldehydes with cyclic silyl enol ethers⁴⁾ in the presence of Lewis acid where the *ortho* trimethylsilyl group in the benzaldehydes (1,2,3) plays a significant role to attain a high selectivity.

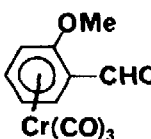
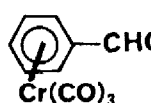
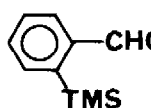
The chromium-complexed benzaldehyde derivatives (1,2,3) were prepared from the parent aldehydes⁵⁾ under the established condition.⁶⁾ The typical experimental procedure for the aldol process is exemplified by the reaction between tricarbonyl(2-trimethylsilylbenzaldehyde)chromium



i: $\text{BF}_3 \cdot \text{OEt}_2 / \text{CH}_2\text{Cl}_2 / -78^\circ\text{C}$ ii: $\text{CAN} / \text{MeOH} / 0^\circ\text{C}$

Table 1. Reaction of Chromium-Complexed and Uncomplexed Benzaldehyde Derivatives with Silyl Enol Ethers

entry	aldehyde	silyl enol ether	aldol product	yield(%) ^a (E : T) ^b
1	1	7	10	61(96:4) ^c
2	1	8	11	70(90:10)
3	1	9	12	75(>98:2)
4	2	7	13	95(93:7)
5	2	8	14	70(97:3)
6	2	9	15	86(>98:2)
7	3	7	16	69(95:5) ^c
8	3	8	17	72(92:8)
9	3	9	18	94(95:5)

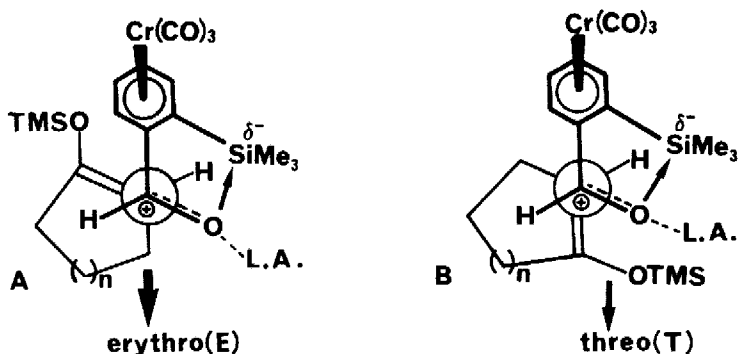
10		4	7	19	84(50:50)
11		5	8	20	91(47:53)
12		6	7	10	63(62:38)

a) Yields of products isolated, after decomplexation, by flash chromatography. b) Determined by 400-MHz ¹H NMR spectra unless otherwise stated. c) Ratio of each pure isomer isolated by flash chromatography.

(1) and trimethylsilyloxycycloheptene (9)(entry 3). To a solution of 1 (58 mg; 0.185 mmol) and 9 (68 mg; 0.37 mmol) in dry CH_2Cl_2 (4 ml) was added dropwise a 1 M solution of $\text{BF}_3 \cdot \text{OEt}_2$ in dry CH_2Cl_2 (0.74 ml; 0.74 mmol) at -78°C under nitrogen atmosphere. After stirring for 15 min, the reaction mixture was quenched by addition of a saturated NH_4Cl solution at the same temperature. Work-up, followed by decomplexation with cerium ammonium trinitrate (CAN)⁷⁾ in methanol at 0°C furnished, after flash chromatography, the erythro derivative (12E; 40 mg) in 75% yield. Careful analysis of ^1H NMR spectrum of 12^{8,9)} indicates that this aldol reaction proceeded in a highly diastereoselective manner.¹⁰⁾ In fact, no threo isomer could virtually be detected in its ^1H NMR spectrum. Other cyclic silyl enol ethers (7 and 8) also showed the similar selectivity. Several results obtained under the standard condition were summarized in Table 1.

A couple of features deserve comment. i) Uncomplexed benzaldehyde derivative (6) gave rise to the erythro derivative in moderate selectivity (entry 12), however the degree of diastereoselectivity is much lower than that of the corresponding chromium complexed aldehyde (1). ii) No characteristic stereoselectivity was observed when tricarbonyl(benzaldehyde)chromium (5) was treated with 8 (entry 11). iii) The ortho methoxy substituent is not adequate enough to attain a high selectivity (entry 10). 4) Changing electron density on the aromatic ring does not affect diastereoselectivity at all.

The mechanism of this highly erythro selective aldol reaction is not manifest yet. However, the above results suggest that both tricarbonylchromium complexation and the ortho trimethylsilyl group are essential for this erythro selective reaction. The benzylic carbocation could be generated by anchimeric assistance of the ortho trimethylsilyl group that can form pentacoordinated species¹¹⁾ and then stabilized by the tricarbonylchromium moiety. This cationic



intermediate^{4b}) should be captured by nucleophilic silyl enol ethers via the transition state **A** rather than **B** affording the erythro product.

Further application of this newly developed aldol procedure is now under active investigation.

References and Notes

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- 8) Stereochemical assignment was made by careful consideration of coupling constant of benzylic proton in each isomer in 400-MHz ¹H NMR spectrum because erythro-threo relative configuration in the aldol reaction can be conveniently assigned from the magnitude of the vicinal coupling constant.⁹) For instance, **10E**: 5.59 ppm, $J=1.5$ Hz; **10T**: 4.91 ppm, $J=9.8$ Hz; **11E**: 5.63 ppm, $J=2.0$ Hz; **11T**: 5.12 ppm, $J=9.0$ Hz; **12E**: 5.38 ppm, $J=2.7$ Hz; **12T**: 5.17 ppm, $J=9.3$ Hz. **12T** was obtained nonselectively by the reaction of **6** with **9**.
- 9) (a) T. Mukaiyama, *Org. React.*, **28**, 203 (1982); (b) D.A. Evans, J.V. Nelson, and T.R. Taber, *Top. Stereochem.*, **13**, 1 (1982); (c) C.H. Heathcock, "Asymmetric Synthesis," Academic Press, New York, **1984**, vol. 3, p 111.
- 10) We isolated chromium-complexed aldol product (**21**), which was converted to **13** after decomplexation. ¹H NMR spectra of **21** and its O-acetyl derivative indicated that **21** was made up of a single isomer with respect to the aromatic ring. Therefore, we would say that this aldol reaction proceeded facial-selectively in regard to the benzene ring of the chromium-complexed benzaldehydes, although the relative stereochemistry is not elucidated yet.
- 11) (a) M. Kira, K. Sato, C. Kabuto, and H. Sakurai, *J. Am. Chem. Soc.*, **111**, 3747 (1989); (b) W.H. Stevenson III, S. Wilson, J.C. Martin, and W.B. Farnham, *ibid.*, **107**, 6340 (1985); (c) R.J.P. Corriu, G.F. Lanneau, and M. Perrot, *Tetrahedron Lett.*, **28**, 3941 (1987); and references cited therein. However, it can't be ruled out the possibility that the aldehyde oxygen directs opposite to the silyl group in the transition state at this stage. Detailed investigation on this point is now in progress.

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